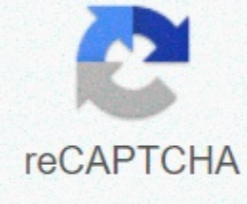




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stable of 110,000 tons per year. Sodium sulfate helps to adjust the level. Unlike alternative sodium chloride, it does not erode the stainless steel vessels used in dyeing. This application in Japan and the United States consumed about 100,000 tons of sodium sulfate in 2006. It is also known as the E514 additive number[22] heat storage capacity to store high heat in phases, transitioning from solid to liquid and temperature change phase at the advantage of 32 °C (90°F), making this material especially suitable for low grade solar thermal storage for later emission in thermal applications in the area. In some applications, the material is included in the heating tiles placed in the attic area, while in other applications. Salt is included in cells surrounded by solar hot water. Phase changes allow the mass of materials needed for efficient heat storage (the heat of the fusion of decahydrate sodium sulfate is 82 kJ/mol or 252 kJ/kg[23]) with the additional advantage of temperature consistency as long as there is sufficient material at the right distance. For cooling applications, the mixture containing general sodium chloride salt (NaCl) reduces the melting point to 18 °C (64 °F), the heat of the fused NaCl - It is more effective, but slow-acting, than its similar magnesium sulfate agent. Sodium sulfate is added to the solution until the crystals are no longer gathered. Two video clips (see above) show that clump crystals when still wet, but some crystals flow freely when the sample dries. It is effective for removing certain drugs, such as paracetamol (acetaminophen), from the body, for example. In 1953, sodium sulfate was proposed for thermal storage in passive solar heating systems. This takes advantage of the unusual melting properties and high heat of crystallization. Other uses for sodium sulfate include frosted windows, flour production as additives in carpet dressings and as additives for cow food, at least one Company Thermaltake makes laptops, computers, cold mats (iXoft Notebook Cooler) using sodium sulfate decahydrate inside padded plastic sheets. The material gradually turns into liquid and circulates, adjusts the laptop temperature evenly and acts as an insulator. Safety, although sodium sulfate is generally considered non-toxic, [22] This risk can be prevented by using eye protection and paper masks. Unlimited transport and no risk phrases or safety phrases. Reference ^ National Biotechnology Information Center [30] Summary of PubChem Compound for CID 24436, Sodium Sulfate. . Accessed Nov 2, 2020. ^Zachariassen, W. H.: Sigler, G. 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