

Bond polarity worksheet



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Name:

Should try to answer the questions without referring to your textbook. If you get stuck, try asking another group for help. Arrange the tapes in each of the following sets in order to increase polarity: C-F, Be-F, O-F, O-Cl, S-Br, C-P, C-S, B-F, N-O Determine based on electronegativity and symmetry if the following molecules are polar or nonpolar: H_2CN CF_4 COCl_2 NF_3 For the following reactions, determine the ΔH°_f based on mortgage dissociation enthalpies: $\text{CH}_4 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_6 + \text{O}_2$ $\text{H}_2\text{N} + \text{H}_2\text{N} \rightarrow \text{C}_2\text{H}_4\text{N}_2$ What is the designation for the hybrid orbital formed from each of the following combinations of atomic orbitals? one s and two p orbitals, one s, three p, and one d For the next molecules, using valence bonding theory, show the orbital overlap of the following molecules. Indicate the hybridization of all orbitals, and the type of overlap used. BCl_3 AlCl_3 CS_2 PF_6^- In order to continue enjoying our website, we ask that you confirm your identity as a human being. Thank you so much for your cooperation. Print test (Only the test content will print) Instructions: The following questions require the use of a reference table containing the electronegativities of different elements.

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