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Ionic formula worksheet answers

28 Formulas for ionic compounds 14-16 In this event, ion formulas help students control cards, consolidate and show their ability to write the right formulas for ionic compounds. Students work collaborating to show each other and the teacher their mastery of ideas. After one group is approved by the teacher, students progressively ven into other groups. Learning goals Students will be able to: write formulas of ionic compounds to remember the names and formulas of common cations and anions. Series of activities Give each student a series of traffic light cards. Ask them to specify whether they know the names of na+, fe2+, no3-, so42-, s2-ions. If they don't know Green, explain what they'll do if they think they know Red: work in groups, check that they can associate the names and formulas of common ions, and show that they can use ion formulas to create formulas for ion formulas of ionic compounds. Give each student a mini whiteboard and ion formula page. Ask them: turn the board upside down to write formulas of whiteboard, sodium bromide, potassium nitrate, calcium sulfate and sodium hydrogencarbonate. Organize students: Ion formulas are in groups of four to complete by typing their leaves in appropriate places. When a group believes that it has successfully completed this task, check its work. If successful, assign them a Group of Experts. This can authorize members of this group to supervise the work of other groups.' Wait for the control cascading. Give successful groups a Formulae for an ion formula card set and an ionic compounds page for each student. Ask them: arrange cards to create formulas of ionic compounds on their page and ask each other questions so you can write formulas of ionic compounds (this can happen in pairs). When a group believes you have competence, check their work using random questions. If five questions are answered correctly, set them as an Expert Group. Authorize members of this group to audit the work of other groups and identify them as expert groups. Wait for the control cascading. With the entire class, ask students to write on their whiteboards: compare sodium bromide, potassium nitrate, calcium sulfate and sodium hydrogencarbonate formulas with what they wrote at the beginning of the session, write an assessment of their progress in Formulae for ionic compounds. Evaluation for learning interpretation Traffic light cards activity gain a personal relevance for students to their learning goals. The two main activities are more about the whole group than on individuals. Students will support and evaluate each other to help the whole group succeed. Personal using mini whiteboards feature. feature start and end increase the confidence that students can develop. Sources Props Formulated sheet Formulated ionic compounds sheet for each student Traffic light cards Mini whiteboard For each group of ion formula cards Ion formula cards Set Formulae Responses to ionic compounds: Compound Formula Compound Formula Sodium carbonate Na2CO3 Zinc nitrate Zn(NO3)2 Silver nitrate AgNO3 Potassium sulfate K2SO4 Calcium bromide CaBr2 Magnesium sulfur MgS Copper hydroxide Cu(OH)2 Aluminum Hydroxide Al(OH)3 Iron(II) Nitrate Fe(NO3)2 Ammonium Chloride AlCl3 Iron(III) Iodide FeI3 Sodium hydrogen carbonate NaHCO3 Lead sulfate PbSO4 Iron(III) carbonate Fe2(CO3)3 Fe2(CO3)3

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