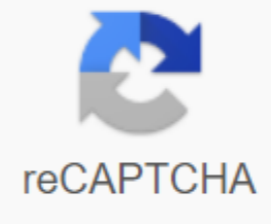




I'm not robot



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Harper college equilibrium lab answers

II. Ammonium System III. Iron thiocyanate System IV. Chromate System V. Nitrogen dioxide system VI. Copper sulphate system In this experiment you are introduced into the chemical balance. You will then be introduced to a number of systems in balance and will be asked to stress these systems by changing the concentration of one of the reactants or products or by changing the temperature of the system. The experiment consists of four parts (Background, Prelab, Experiment, Postlab) that should be completed in the order listed below. Introduction Background Prelab Experiment postlab Day 18: Le Chatelier's Principle (continued) Select the Experiment section. Read the information in An example to learn the icons that are used throughout the lab. Complete The Le Chatelier's Principle Online Lab. (File and answer button below) Note: Some systems have special comments or notes with information needed to understand specific voltages. [le_chatelier_online_lab_assignment_2020.docx](#) File size: 28 kb File type: docx Download File [le_chateliers_principle_online_answers.pdf](#) File size: 898 kb File type: pdf Download File Day 19: The Haber Process Watch the following videos: Complete the The Haber Process (file and answer key below) [the_haber_process_assignment.doc](#) File Size: 27 kb File Type: doc Download File [the_haber_process_assignment_answers_2020.doc](#) File Size: 39 kb File Type: doc Download File : Le Chatelier's Principle Graphing Watch the following video: Fill in the Equilibrium: Graphing notes. (File and answer key below) [equilibrium_graphs_no_notes.docx](#) File size: 125 kb File type: docx Download File [le_chatelier_graphing_answers.pdf](#) File size: 425 kb File type: pdf Download File Download Download The worksheet Interpreting Graphs Fill in. (File below) [le_chatelier_graphing_ws.pdf](#) File size: 88 kb File type: pdf Download File Day 21: Le Chatelier's Principle Wrap-Up In Unit 8- Color Change Reactions: Exploring Equilibrium in the Open Learning Initiative: Complete Module 27: Iron (III) Solutions: Watch the Introduction to a New System video (see 241). Read and complete Introduction to Iron (III) Thiocyanate (p. 242). Read and complete Iron (III) Thiocyanate Sort (p. 243). Read and complete Iron (III) Thiocyanate Molecular Sort (p. 244). Fill in the questions adding iron prediction before the laboratory (p.245). Complete the Virtual Lab: Add Iron Test (p.246). For directions, click on the Light Bulb icon below the lab screen. Fill in the post-lab questions to add iron explanation (p.247). Complete the Virtual Lab: Endothermic vs. Exothermic (p.248). For directions, click on the Light Bulb icon below the lab screen. Fill in the post-lab question (p.249). Complete the post-lab question Endothermic vs. Exothermic Explanation (p.250). Watch the video Conclusion (p. 252). Challenge! Complete the Exploring Exploring Reflection and evaluation (p.251). I will give you feedback on your written answers when you complete the Refection Questions and Quiz. Day 22: Online Homework (Quest Learning & Assessment) Complete the online homework assignment Equilibrium Quick Check #3. End of balance week 5... continue to Equilibrium week 6. 6.