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Match each term with its definition

Metro (apocope of metropolitan trains) is a system of trains in the city located in the city and its vicinity. It is characterized by the large transport of passengers in major cities, combining their diverse and environmental zones, with high capacity and frequency and in isolated forms to other systems of transportation. Metropolitan trains are built underground or on the surface, although most systems will use a mixed model that sections are combined in both modalities 1 - as a result, metro tends to reduce the efficiency of car travel for two reasons firstly due to the transfer of more users from the car and second due to the reduction of the purchase of cars in the city center, the use of the integrated sector of the metro. Match each of the words below with a definition or description. 1 point in titration when the added number of standard reagents equals the amount of titration analysis. 11) End point 2 analysis is titrated with standard reagent and the volume of standard solution needed to make the reaction as a measure of 12) indicator 3 pure and stable reagent which can be used directly after weighing 13) Direct titration 4 analysis with analysis in titration of 14) Reverse titration 5 standard reagent is added in excess to ensure that the complete reaction to the analysis of excess reagent is titrated with the second standard substance 15) indirect titration 6 solution, With the concentration known, usually made from the chemical of the known purity of 16) standard 7 standard reagent of the known concentration is added from the buret to the analysis solution of 17) Standard Solution 8 analyte does not respond directly to the titrant so it is converted to another form which will respond to titrant 18) titrant 9 points in titration when there is a change in problem solving analysis is observed, demonstrating parity. It was added to solve the problem and help to observe the success of the reaction 20) match each word with its definition or description, \_\_\_\_\_The the point in titration when the added number of standard reagents equals the amount of titration analysis, a volumetric analysis technique used to identify unknown concentrations of the solution using the known concentration of another unknown solution\_\_\_\_\_. A standard substance of a known concentration added by buret to solve the analysis problem \_\_\_\_\_. A common concentration unit - mole per liter, titration point when an analytical solution changes are observed, indicating the appropriateness of \_\_\_\_\_. The solution of the unknown concentration \_\_\_\_\_ is a standard solution of a known concentration that is added from the burate to the analytical solution \_\_\_\_\_. A common concentration unit - mole per liter, titration point when an analytical solution changes are observed, indicating the appropriateness of \_\_\_\_\_. At the end of the point, the point at the end of the titration point, the point of equal point, the bank analyzes the titration, the titration indicator, the drag corresponding to each word with the definition. To respond, drag the appropriate word from the left column to the definition on the right. Each word may be used more than once or not at all. Each correct match is worth one point. Note: Each valid selection has one value. Point