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Chemistry chapter 4 assessment answers

1. The normal chemical reaction present in the protein does not change during. Just move external electrons. Positive charges form when electrons are lost. 3. P, I, Cl, and O will configure the ananas because they are non-metals. Magha, I, Cs, B, and the company will configure because they are metals. 5. (a) P3- (b) Mg2 s s (g) Al3 s s (d) O2-; (e) Cl-; (f) Cs + 7. (a) [Ar] 4s23d104p6; (b) [Kr] 4d105s25p6 (c) 1s2 (d) [Kr] 4d10; (e) [He] 2s22p6; (F) [Ar] 3d10; (g) 1s2 (h) [they] 2s22p6 (i) [Kr] 4d105s2 (j) [ar] 3d7 (k) [ar] 3D 6, (l) [ar]. (a) 1s22s22p63s23p1; Al3 +: 1s22s22p6; (b) 1s22s22p63s23p63d104s24p5; 1s22s22p63s23p63d104s24p6; (g) 1s22s22p63s23p63s23p63d104s24p65s2 sr2 sr2 sr2s22p63s23p63d104s24p6; (1 (ڈی22s1 1 :+ لی ؛s2 1 (ای) ؛s22s22p63s23p63d104s24p3; 1s22s22p63s23p63d104s24p6; (f) 1s22s22p63s23p4; 1s22s22p63s23p6 11. NaCl contains the giant ions arranged in a crystal-swell, not the co-continental bandwa inn. 13. ionic: (b), (d), (g), and (i); Co-a, (a), (a), (f), (h), and (k) 15. (A) Cl . (d) S; (e) N . (f) P (g) N 17. (A) H, C, N, A, F. (b) H, I, Br, Cl, F; (h) H, P, S, A, F. (D) Na, l, h, a. (e) Ba, H, N, A21. (A) HF , (b) Company (j) Oh , (D) Pkl , (e) Paul; (f) PO s (G) CN 23. (A) Cesium chloride s (b) barium oxide slfide slfide slfied sl (d) berylum chloride sl (e) hydrogen bromide s (f) aluminum fluoride 25. (A) RbBr (b) MgSe s (g) Na2O s (D) CaCl2 s (E) HF; (f) Rehna s (g) AlBr3 s (h) (NH4) 2SO4 27. (A) CLO2 s (b) N2O4 s (g) K3P s (D) Ag2S s s s (E) AIF3 . 3H2O s (f) SIO2 29. (a) Cromme (III) Oxide s (b) Iron (II) Chloride; (g) cromme (VI) oxide s (d) titinion (IV) chloride s (e) cobalt (II) chloride headrati; (F) M.B.B.D. (IV) Sulfide 31. (A) K3PO4 s (b) CuSO4 s (g) CaCl2 s (d) TiO2 s (E) NH4NO3; (f) NaHSO4 33. (a) Managing (IV) (b) Mercury (I) Chloride (J) Iron (III) Nitret , (d) Titinum (IV) Chloride s (E) Cooper (II) Bromide 34. (a) Eight electrons: (b) No electrons Be2 s (d) Eight electrons: (e) no electrons Ga3 s (f) No electrons Li s (g) Eight electrons: 36. (a) (b) (c) (d) (f) 38. 40. (a) In this case, the Lewis structure is insufficient to describe the fact that experimental studies have shown two unpaired electrons in each oxygen inn. (b) (b) (b)