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The chapter explains how intermolecular forces work between particles of matter. It also explains that these forces differ from the pure electrostatic forces that exist between two opposite-charged ions. In addition, they do not include the forces that hold the atoms of the covalent molecule together. In addition, intermolecular interactions determine the state of matter and the chemical properties of the substance, which do not change with the change of state. Reactivity depends on the physical condition. On the other hand, Charles' law and Boyle's law are also explained in this chapter. It is noteworthy that the forces of interaction between gas molecules are practically independent of their chemical nature and insignificant. There is an equation of the state of the ideal gas. At low temperature and high pressure, intermolecular forces begin to work strongly between gas molecules. In addition, under suitable conditions such as temperature and pressure, gases can be drained. Topics covered by the State of Matter Look at sub-topics that are covered in Matter States.The behavior of real gases - Deviations from ideal behavior - This topic is developed by Boyle's law, and the application of the ideal gas equation. Gas Laws - This topic explains the different gas laws. Perfect Gas Equations - This Subtheme Gives Us ideal gas equations and equations of other gas laws. Intermolecular Forces - This subject explains the various intermolecular forces. Kinetic Molecular Gas Theory - This subject to the theme teaches kinetic molecular theory of gases and postulates of it. Liquefaction of gases - This subject explains the liquefaction of different gases. Gaseous State - This subtem explains the gas substance and its characteristics. Fluid condition - This subject explains liquid substances and their characteristics. The importance of Notes to ReviseRevision Notes are important and very easy to remember for their easy-to-understand format. In addition, we do not need to open a few books until the day or two before the exam, as all items are covered by these revision notes. The key points of all topics are covered in revision notes, so students can rely on these notes to prepare for the exam. Review notes help to increase students' confidence as they can review all topics covered in the curriculum in a short time. Both grades 11 and 12 form the steps in a student's career. Because grades in grade 11 and grade 12 exams decide the student's future. Thus, students should focus on their studies and study and understand all the basics as clearly as possible. In addition, in order to crack medical or engineering entrance exams, a good knowledge of subjects proves useful. Note to review Features Features Feature notes on Vedantu cover the following types of questions. Brief answers to questionsVer short answers to questionsPoints to rememberSver's key notes can be passed on to students for their last-minute review. 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