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Smog smog photochemical reactions are a term originally formed by a combination of words smoke and fog. This may seem a little confusing, because Los Angeles, infamous for its smog, is hardly known for foggy weather. Confusion arises because there are actually two types of smog: London smog (well know for its fog) and Los Angeles smog (also known as photochemical smog). Photochemical smog is controlled by u.v. energy from the sun, and Los Angeles is better known for its sunny weather. The differences between the two types of smog are summarized in the table below. Title: London smog (New York smog, gray smog) Photochemical smog (Los Angeles smog, Denver smog, brown smog) Weather: cool, humid solar content: particles, NOx sulfur oxides, ozone, hydrocarbons. Sources: coal, etc. gasoline, burning. On a smoky day, there are literally thousands of reactions that occur in the atmosphere. Fortunately, there are a few that can help us initially understand the formation of photochemical smog. In the reactions listed below, the most important constituents are described with a larger, bolder font. 1. **NO** - O₂ → and u.v. → NO₂ In this first reaction we start with nitric oxide (NO),

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