

Module 3 – Nautical Science Unit 5 – Physical Science Chapter 20 - Basic Electricity Section 1 – What Is Electricity?



What You Will Learn to Do

Demonstrate an understanding of Physical Science



- 1. Describe the fundamental theory of electricity
- 2. Describe the properties of conductors and insulators
- 3. Describe the six common methods of producing voltage







CPS Key Term Questions 1 - 8





Electricity -

The presence and motion of electrons, protons and other charged particle; manifests itself as attraction, repulsion, luminous and heating effects

Battery -

One or more cells assembled in a common container to act as a source of electrical power





Conductor -

Insulator -

Schematic -

A material or object that permits an electric current to flow easily

Materials that are poor conductors (as in electricity or heat); Materials that have few free electrons

Of, relating to or in the form of a scheme or diagram



Key Terms

Electromotive force -> Symbol: E or emf

Photoelectric -

Voltage that causes current to flow through a circuit; the potential difference in charge between two points in a circuit (as in a cell or generator)

Involving, relating to, or utilizing any of various electrical effects due to the interaction of radiation (such as light) with matter





Thermocouple -

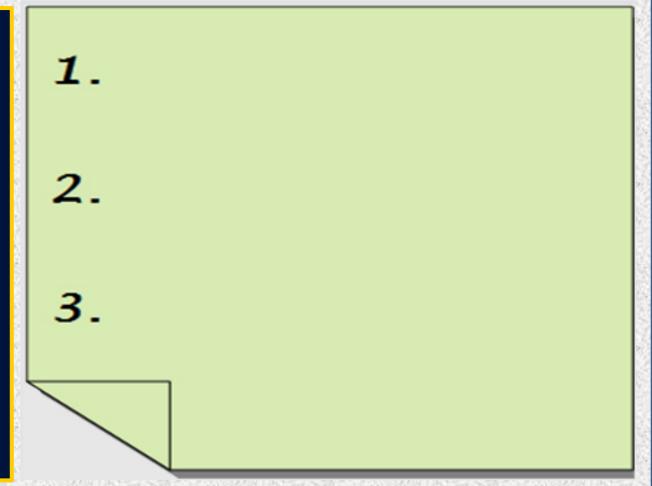
A device for measuring temperature in which a pair of wires of dissimilar metals (such as copper and iron) are joined and the free ends of the wires are connected to an instrument (such as a voltmeter) that measures the difference in potential created at the junction of the two metals



Opening Question



Name the three best conductors of electricity, and explain why two of them are used most often.



(Use CPS "Pick a Student" for this question.)

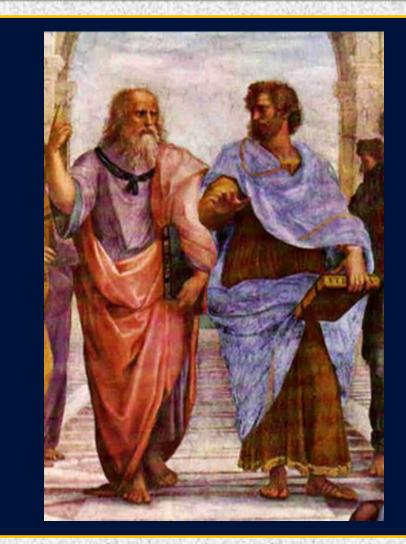


Warm Up Questions



CPS Lesson Questions 1 - 2





The study of electricity began with the ancient Greeks.



Rubbing amber with a cloth created a force that attracted the cloth to the amber.





Rubbing two pieces of amber with two cloths caused the cloths to repel one another as much as they were attracted to the amber.





The forces the Greeks observed were called electric (from the Greek word for amber).

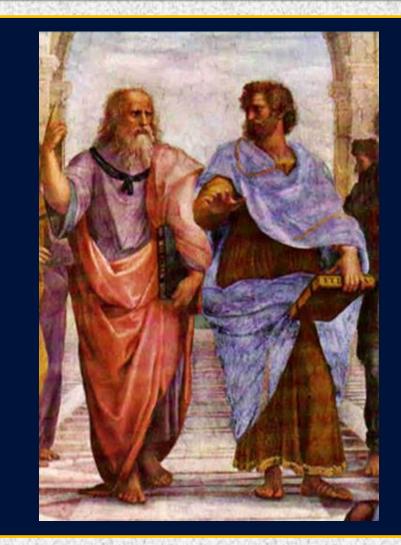




The cloths and amber were said to be electrically charged.



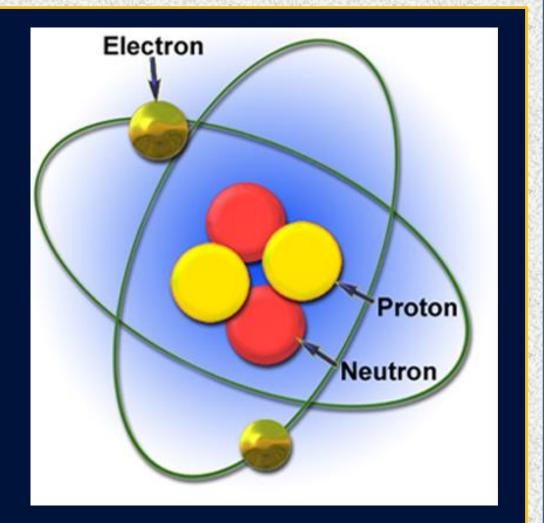




The Greeks could not explain electrical force.

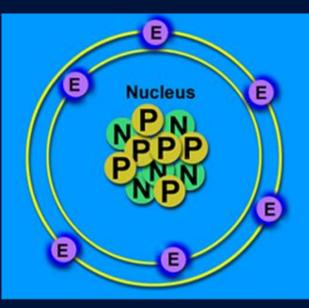


The true cause of electricity was determined with the development of the atomic theory of matter.





Scientists could explain electrical charges when they found atoms were composed of negatively charged particles (electrons) orbiting positively charged particles (protons) and neutrons which have no charge.





ATOMS AND ELECTRICITY

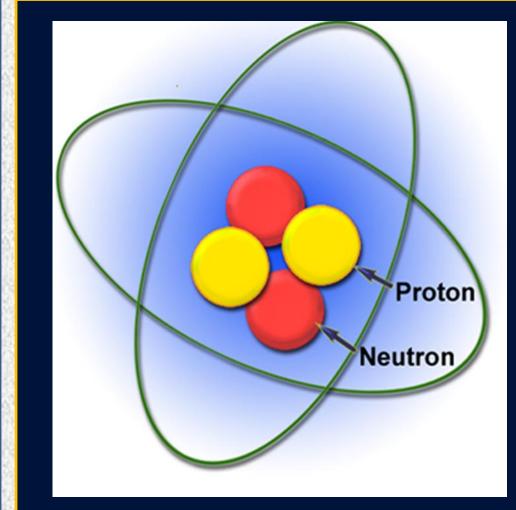
Atoms, such as the Lithium atom below, are composed of a nucleus...



Under most conditions, an atom will have no charge.





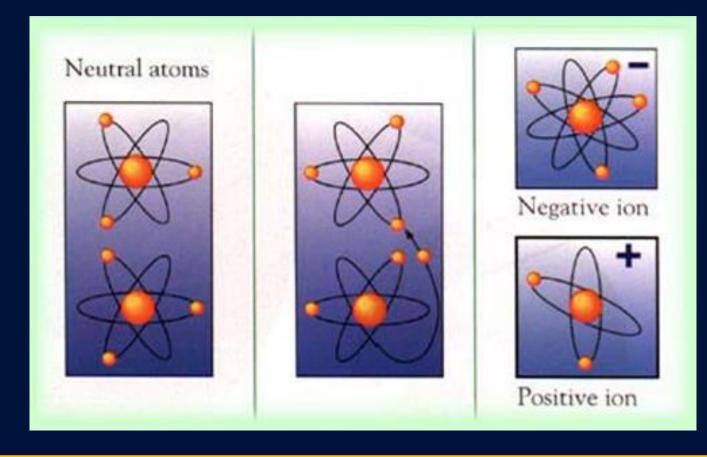


If the number of electrons is increased, an atom becomes negatively charged.

If electrons are removed, an atom will have a positive charge.

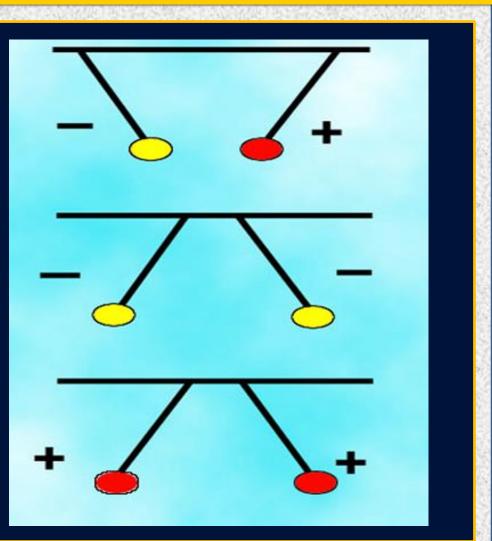


Charged atoms are called ions.



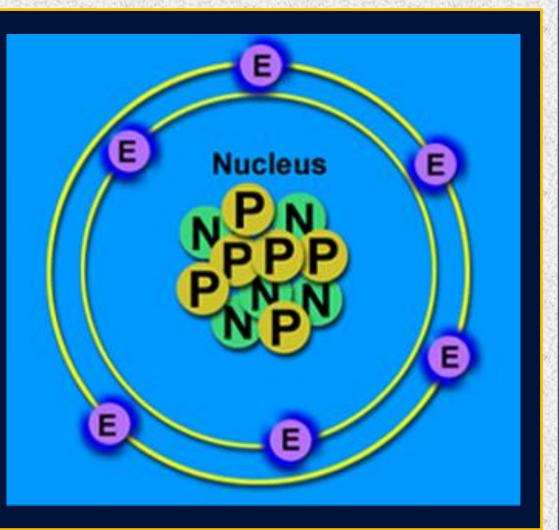


Unlike charges attract each other while like charges repel each other.





In the atom, electrons are held in their orbit by the attractive force between them and protons in the nucleus.





In the Greeks' experiments with amber, the cloth picked up electrons from the amber, becoming negatively charged.

This left the amber with a positive charge, and unlike charges attract one another.





Check On Learning Questions



CPS Lesson Questions 3 - 4



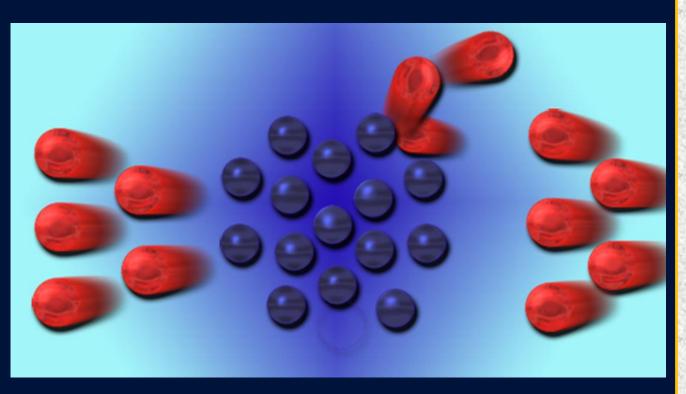


Conductor

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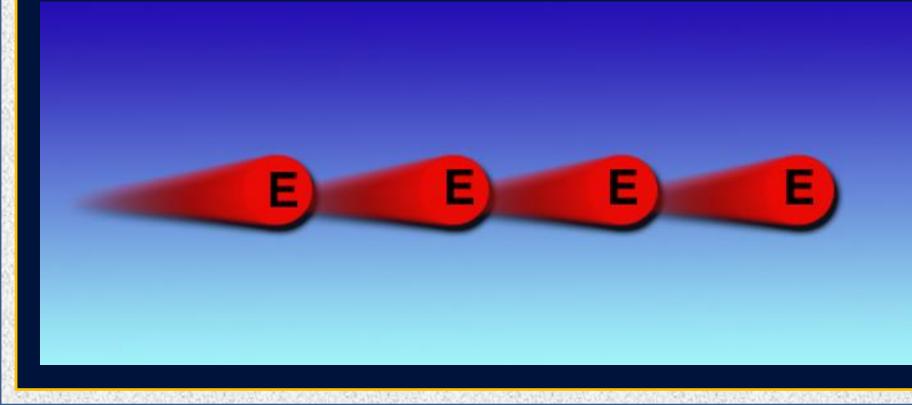
An electric charge can move through a material if it has a large number of free electrons. **Electrons** can



easily move from atom to atom in material with a large number of free electrons.



Substances that allow free movement of electrons due to their atomic structure are called conductors.







Silver, copper, and aluminum wire, in that order, are the best conductors.





Copper and aluminum wire are the most commonly used because they are the least expensive.

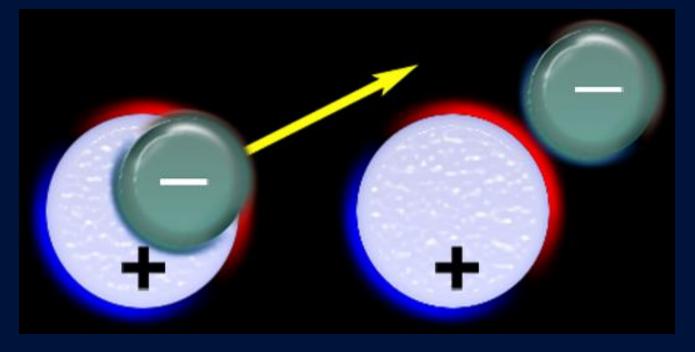


Electrical energy is conveyed as a wave traveling at the speed of light through conductors by free electrons.



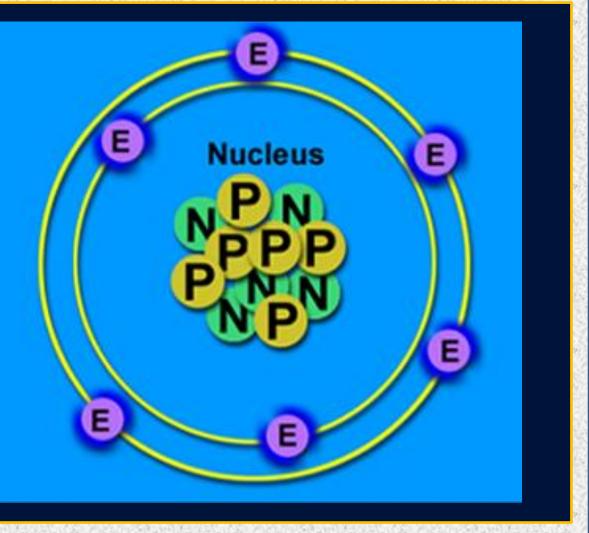


As the electrical energy passes, each electron moves a short distance to the next atom, displacing one or more electrons by forcing them out of their orbits.

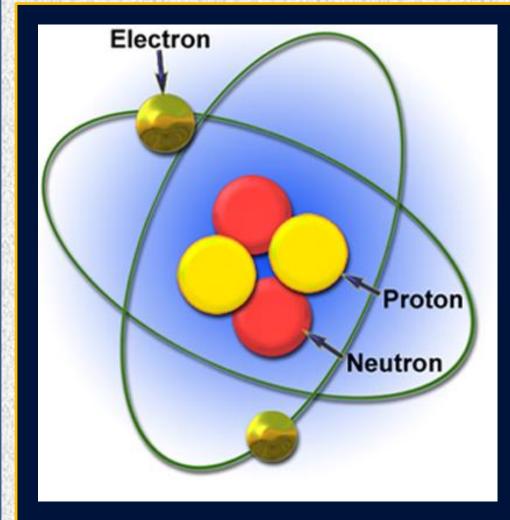




The replaced electrons repeat the process in other nearby atoms.







Some substances have very few free electrons and are therefore poor conductors.



These substances, such as rubber, glass, or dry wood, are called insulators.







Wood

Glass



Conductors and Insulators

Good conductors such as wire carry electricity and are covered by insulating material to prevent electricity from being diverted from the conductors.

Insulator

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Check On Learning Questions



CPS Lesson Questions 5 - 6

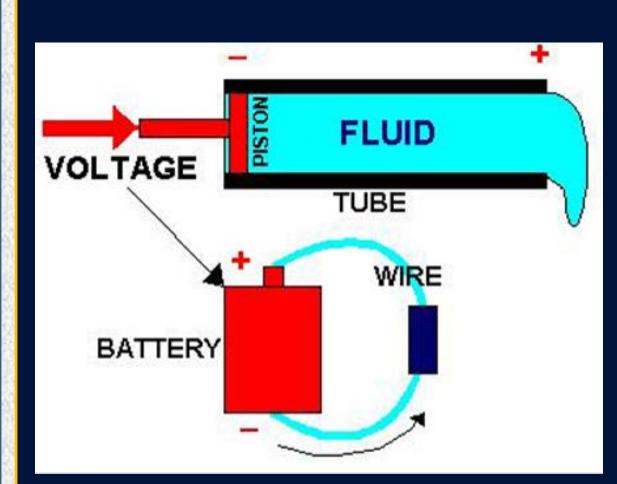












The force that causes electricity to move in a conductor is called voltage or electromotive force, symbolized by the letter *E*.

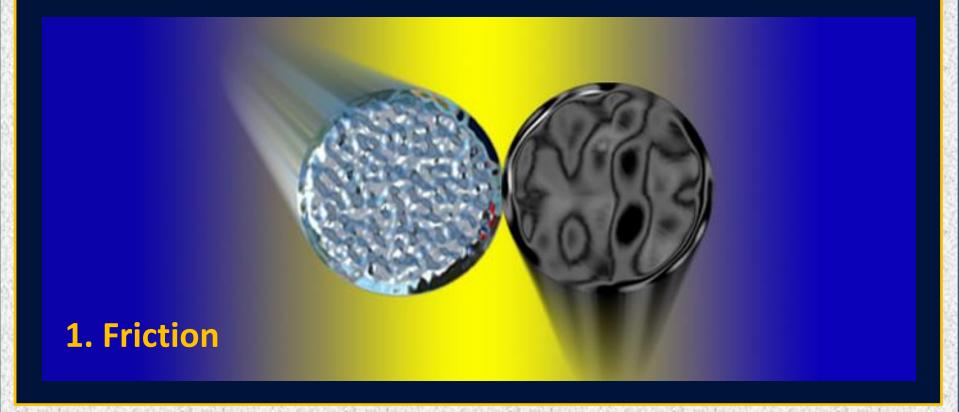


There are six basic ways to generate voltage.

- 1. Friction
- 2. Pressure
- 3. Heat
- 4. Light
- 5. Chemical action
- 6. Magnetism



Voltage can be produced by rubbing two materials together.









Static electricity is the most common name for electricity generated through friction.

It occurs often in dry environments.

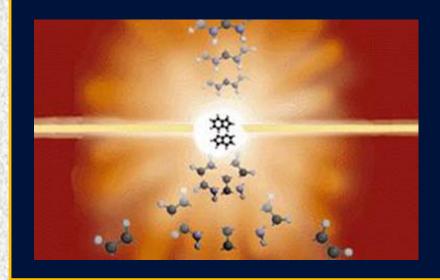




2. Pressure:

Creating pressure by squeezing crystals, such as natural quartz or manufactured crystals, produces voltage.





Compressed electrons tend to move through a crystal at predictable frequencies.



Crystals are frequently used in communications equipment.

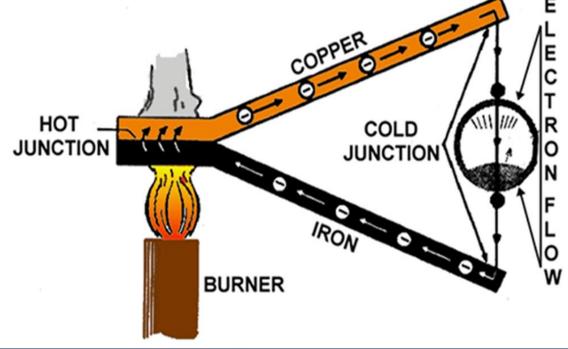




3. Heat

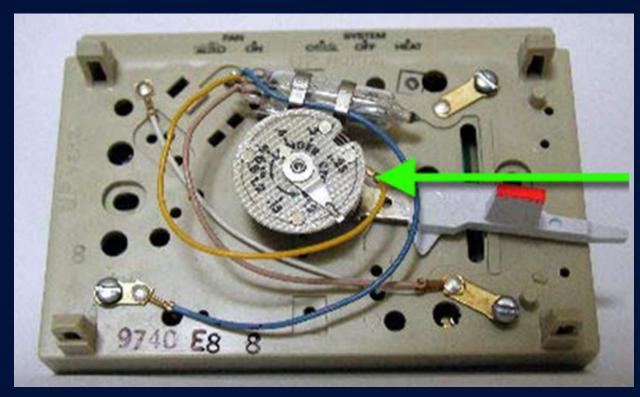
Voltage can be produced by heating the place where two unlike metals are joined.

The difference in temperature of the two metals determines the amount of voltage.





The hot junction where the moving electrons from the metals meet is called a thermocouple.









Thermocouples are often used to measure and regulate temperature, as in a thermostat.

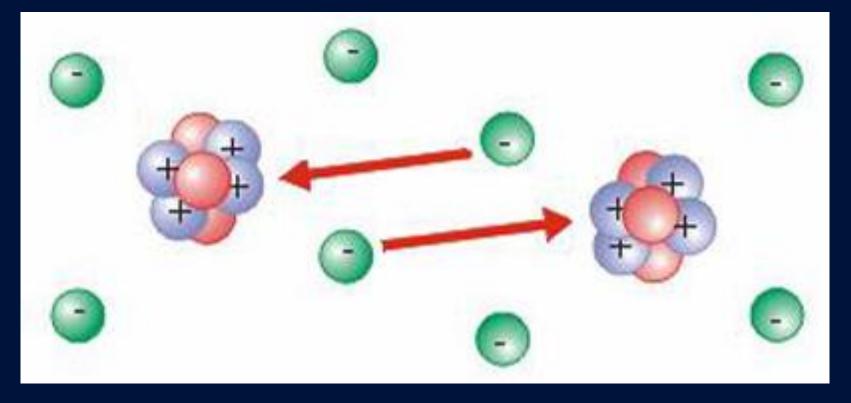


Photoelectric voltage is produced when light strikes a photosensitive (light-sensitive) substance.





Light dislodges electrons from their orbits around surface atoms.





The photoelectric cell operates on this principle.





Use of light-produced voltage is growing. This type of electricity can be produced without generating carbon-based waste products or depleting natural resources.

Applications include:

- buildings
- batteries
- cameras
- burglar alarms





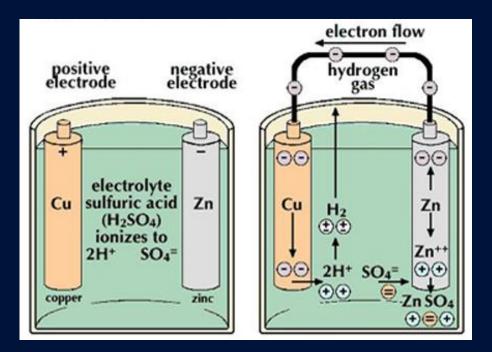
Voltage can be produced by chemical reactions, as in a battery cell.

5. Chemical action



A simple voltaic battery consists of a carbon strip (positive) and a zinc strip (negative) suspended in a solution of water and sulfuric acid.

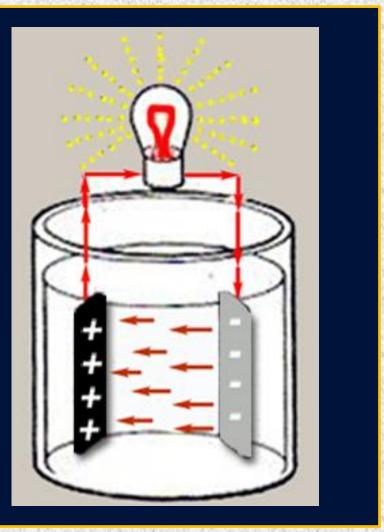
The solution is called the *electrolyte*.





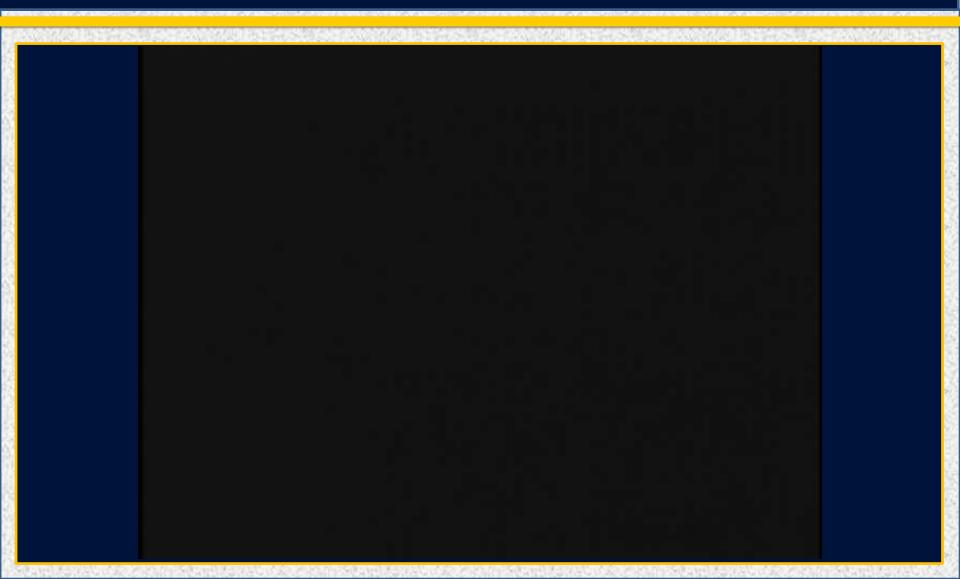


The chemical action that results from this combination causes electrons to flow between the zinc and carbon electrodes.



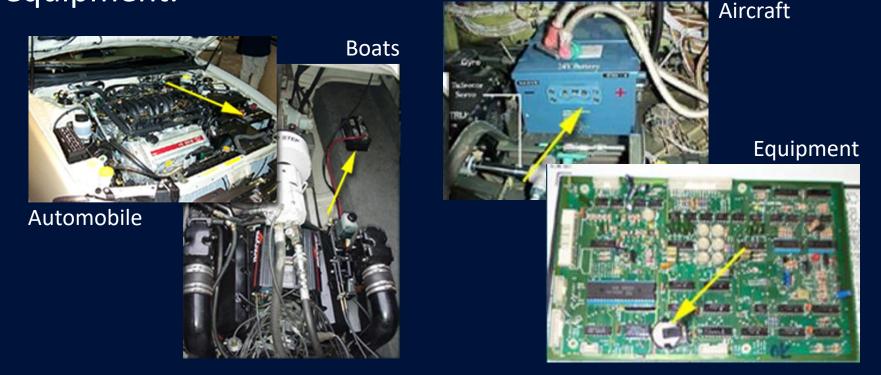




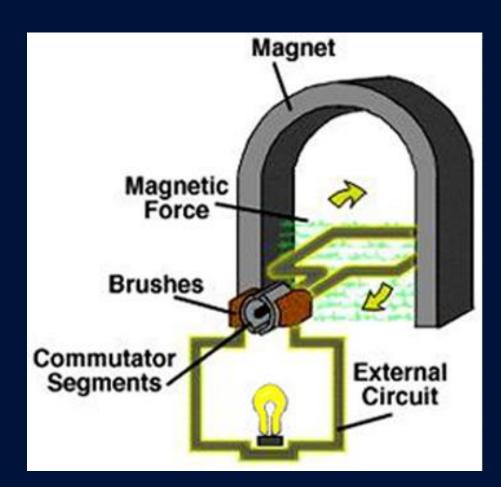




Batteries are used as sources of electrical energy in automobiles, boats, aircraft, ships, and portable equipment.







6. Magnetism

Voltage can be produced when a conductor moves through a magnetic field cutting the field's line of force.

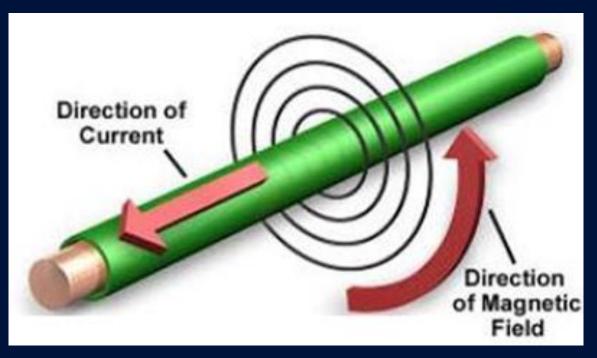


This method is used in <u>electrical generators</u> and is the most common source of power.





Usually, a copper-wire conductor is moved back and forth through the magnetic field created by a U- or C-shaped electromagnet.







An instrument designed to measure voltage in an electrical circuit is called a voltmeter.

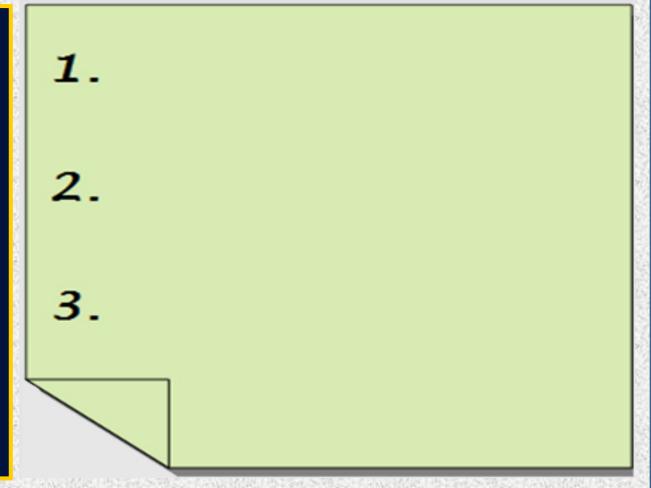




Review Question

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Name one mechanism using each of the following voltage sources: pressure, light, chemical action, and magnetism.



(Use CPS "Pick a Student" for this question.)



Closing Questions



CPS Lesson Questions 7 - 8



Questions?

